

AZIZ FATIMAH MEDICAL & DENTAL COLLEGE



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BIOCHEMISTRY DEPARTMENT IN A GLANCE

Biochemistry is the dynamic, exciting science in which chemistry is applied to the study of the atoms and molecules which comprise living organisms. This includes organic molecules

and their chemical reactions. It has revolutionized our understanding of and provides a backbone to modern medicine.

Biochemistry Department at Aziz Fatimah Medical & Dental College has a unique approach to the biochemical sciences that cultivates critical thinking as well as depth of knowledge by exposing its students to the full spectrum of modern biochemistry. The comprehensive teaching and assessment plan is strategically designed according to the UHS and PMC syllabi and guidelines to achieve maximum results.

The strength of Biochemistry Department is its conductive environment and committed staff. The vibrant teaching staff is highly qualified with post graduates degrees and certifications along with vast teaching experience. The department's aim is establishment of research culture and encouragement of student participation in it.

Biochemistry department has a well equipped laboratory and is managed by qualified and experienced technical staff. Practical training of the subject has been devised and the focus has been shifted to diagnostic biochemistry, in accordance with the UHS syllabi.

The department follows a proficient and result oriented teaching and assessment plan which includes new and interesting teaching strategies. Learning is made easy by increasing interactive student teacher sessions. Students are evaluated in cognitive, psychomotor and applied domains by conduction of regular formative and summative assessments like multiple choice questions, quizzes, written tests, assignments, presentations and OSPE and oral viva. At the end of each academic year a university standard send-up examination is conducted.

Biochemistry Department Team-AFMDC

Positions	Name
Head of Department	Prof. Dr. Shakeel Ahmad
Associate Professor	Dr. Saira Saad
Assistant Professor	Dr. Sajjad Ghani
	Dr. Jameel Ahmad
Demonstrators	Dr. Huda
	Dr. Mehwish
	Dr. Urwah
Laboratory Incharge	Miss Qurat-ul-Ain
Laboratory Assistant	Mr. Saleem
Computer Operator	Mr. Ramzan

Assessments



MONTHLY ASSESSMENTS

> SEND UP EXAM

LEARNING OBJECTIVES OF BIOCHEMISTRY

TOPIC	SUBTOPI	LEARNING OBJECTIVES	LECTURE	SGD	PRACTICA
	CS				L
	Monosacc	• Define and classify carbohydrates.	5		Qualitative
	harides	• Discuss isomerism in			analysis of
		monosaccharides.	_		carbohydrate
RY		 Discuss chemical and physical 			S
LSI		properties of carbohydrates.	_		
M		Discuss monosaccharides of			
		biochemical importance.			
ы	Disacchari	• Discuss disaccharides of biochemical	1		Qualitative
AT	des	importance.			analysis of
DR.		• Discuss oligosaccharides of	1		carbonydrate
IXI		biochemical importance.	4		5
OE	Polysaccha	 Discuss homopolysaccharides of 	1		
RB	riues	Diochemical importance.	1		-
CA		 Discuss heteropolysaccharides of highering limportance 	1		
	Fatty agida	Uighlight Dielegiest significance of	1		-
	ratty actus	• Highlight Biological significance of	1		
		 Classify lipids 			
		 Discuss fatty acids 	1		Qualitative
	Triacyloly	 Discuss fully delds. Discuss structure of TAGs 	2		analysis of
	cerols	 Identify chemical and physical 	-		carbohydrate
	•••••	properties of TAGs			S
		 Discuss lipid peroxidation 			
	Phospholi	Discuss structure, properties and	1		-
	pids	significance of phospholipids.			
	-	• Describe lung surfactant, platelet			
		activating factor and cardiolipin.			
X		• Identify enzymes involved in			
1ISTR		degradation of phospholipids.			
	Glycolipid	• Enlist types of glycolipids along with	1		
IEV	S	their significance and degradation.			
CH	Eicosanoid	• Discuss origin, half life, potency,	1		Qualitative
A	S	functions and clinical significance of			analysis of
I		prostaglandins, thromboxanes and			carbohydrate
		leukotrienes.			S

	Cholestero l	• Describe the role and properties of cholesterol and its related compounds (bile acids).	1	
E CHEMISTRY	Nucleotide s	 Define nucleotides and nucleosides. Enumerate purine and pyrimidines along with their nucleosides and nucleotides. Identify biomedical importance of nucleotides. Discuss unusual bases. Explain role of bases as drugs. Discuss role of ATP. Discuss biochemical functions of nucleotides and cyclic nucleotides. 	2	
JCLEOTID	Nucleic acids	 Identify structural features of DNA. Identify types and structural features 	2	Qualitative analysis of carbohydrate s
N		of RNA.		
		UQs Discussion	1	
	Types	• Give organization and composition of eukaryotic and prokaryotic cells.	2	
	Cell membrane	 Explain biochemical composition of cell membrane. Discuss membrane asymmetry, glycocalyx, blood group antigens. Identify importance of cholesterol in membranes. 	1	Qualitative analysis of carbohydrate s
TRANSDUCTION	Transport across cell membrane	 Describe diffusion (simple and facilitated), osmosis and osmotic pressure, transport of charged molecules and Gibbs-Donnan equilibrium. Outline Pores (aquaporins), channels and carriers. 	2	
CELL & SIGNAL	Types	 Discuss types of signals. Define gap junctions, autocrine, paracrine and endocrine signals. 	1	Qualitative analysis of

Receptors	 Enlist various types of receptors. Explain ligand gated ion channels, G-protein coupled receptors, Catalytic receptors and intracellular receptors. Discuss receptor tyrosine kinases. Classify G-proteins. Qualitative analysis of carbohydrates 	2	carbohydrate s
Second messenger s	 Elaborate Adenylyl cyclase and cAMP cascade. Explain Phospholipase and IP3, DAG cascade. Discuss calcium calmodulin cascade. 	2	-

TOPI C	SUBTOPI C	LEARNING OBJECTIVES	LECTURE	SGD	PRACTIC AL
	Water	 Discuss the biomedical importance of water. Identify water molecule as a dipole. Discuss hydrogen bonding. Explain properties of water as a solvent. 	1	1	Introdoucti on to use of laboratory facailities / equipment including safety
TER AND PH	рН	 Define weak & strong acids and bases. Discuss ionization of water. Define pH and pKa Explain Henderson-Hasselbalch equation and its applications Discuss titration curves of a weak acid and an amino acid. Define zwitter ion and isoelectric pH. 	2	-	measures
	Buffers	 Define buffers. Elaborate bicarbonate and Hb buffer system. Discuss transport of CO2 and chloride shift. Elaborate phosphate and protein 	2	1	• Sol
	•	buffer system.			utio
WA	Techniqu es	• Discuss pH metry & isoelectric focusing	1		ns ⪻

	Amino acids	Define Amino acid.Discuss optical forms of amino acids.	2	1		epar atio
		• Identify the biomedical importance of amino acids.				n of solu
		Classify amino acids.				tion
		• Discuss physical and chemical properties of amino acids.	-			S
	Proteins	Classify proteins.	5			
		Discuss peptide bond.			•	Intr
		• Discuss primary & secondary structure of protein.				odu ctio
		Discuss tertiary structure of protein.	-		-	n and
		• Elaborate protein folding.	_		-	con
		• Identify protein misfolding diseases			_	vers
	Plasma	• Discuss structure, types and	4			ion
	proteins	biomedical significance of plasma				of
		proteins.				con vent
						iona
						1
						and
						SI mea
RY						suri
LLS						ng
IW						unit
CHI	Protein	Chromatography	3		•	s. Test
NI	separatio	Electrophoresis			-	s to
ETC	n	Centrifugation	-		-	dete
PR(technique	-				ct
	S Collagen	• Identify the components of	2		-	prot eins
IX	conagen	extracellular matrix.	-			/
TR		• Enlist types of collagen.				pept
MA		• Discuss the structure of collagen.				ides
AR		• Identify the steps in synthesis of				/ ami
ELLULA		collagen.	_		-	no
		Discuss collagen degradation.Discuss collagenopathies.				acid s
RAC	Elastin &	• Discuss the structure and synthesis	2		Tests to	0
KTR	Keratin	and diseases of elastin.			detect	
E					protein	is /

		 Identify the role of α1-antitrypsin in elastin degradation Discuss keratin. 		peptides / amino acids
	Structura l proteogly cans	Discuss Fibronectin and LamininDiscuss proteoglycans.	2	
	GAGs	 Discuss the structure and functions of glycosaminoglycans GAGs . Discuss muco-polysaccharidosis. 	2	
	Porphyri ns	 Identify the structural features of porphyrins. Enlist haem proteins. Identify the site of haem synthesis. Discuss rate limiting step of haem synthesis along with its regulation. 	2	Tests to detect proteins / peptides / amino acids
JLISM & PORPHYRIN	Haem	 Discuss rate limiting step of haem synthesis along/with its regulation. Elaborate the haem synthetic pathway. Discuss porphyrias. Identify haem degradative pathway. Discuss hyperbilirubinemias causing iaundice 	4	
HEMOGLOBIN, HEME METAB	Hemoglo bin	 Discuss structure of myoglobin and hemoglobin. Discuss various types of hemoglobin. Elaborate the binding of oxygen to hemoglobin. Discuss Factors Effecting Hb-Oxygen Binding. Discuss sickle cell anemia. Discuss various types of hemoglobinopathies. 	5	Tests to detect proteins / peptides / amino acids
Ŧ		UQs Discussion	1	

TOPIC	SUBTOPI	LEARNING OBJECTIVES	LECTURE	SGD	PRACTICA
	С				L

	Intro	 Define enzyme and substrate Discuss the biomedical importance of enzymes Identify the nature and structure of enzymes. Classify enzymes. 	1	1	Qualitative analysis of lipids
	MOA	• Discuss the mechanism of action of enzymes.	1		
	Enzyme action	 Discuss the properties of enzymes. Discuss Michealis –Menten Equation. Identify the factors effecting enzymes action 	1		
	Regulation	 Discuss Line- Weaver Burke Plot. Explain effect of inhibitors on enzyme action. Discuss regulation of enzymes. 	1		
IZYMES	Isoenzyme s	 Discuss regulation of enzymes. Define iso enzymes. Discuss the clinical significance of iso enzymes. Identify the therapeutic uses of enzymes. 	1	1	Qualitative analysis of lipids
EN		UQs discussion	1		
	Introducti on	 Give definition, classification and requirement for humans of various vitamins. Enlist factors affecting the vitamin content of food. 	1		
	Water soluble vitamins	• Discuss important dietary sources, RDA, intestinal absorption, transport, storage and diseases associated with water soluble vitamins	5	1	Physical & chemical analysis of urine to detect normal constituents.
VITAMINS	Fat soluble vitamins	• Discuss important dietary sources, RDA, intestinal absorption, transport,storage and diseases associated with fat soluble vitamins.	5	1	Physical and chemical analysis of urine to detect normal constituents.

Macro minerals	 Give introduction to minerals and trace elements. Discuss Calcium and phosphorus metabolism. Discuss Wilson disease, tetany, hypercalcemia. Discuss Phosphorus, magnesium and sulfur. 	1 2 1	1	Collection and storage of urine samples for laboratory analysis, and physical and chemical analysis of urine to detect abnormal constituents.
Electroly	e • Discuss Sodium, potassium and chloride.	3	1	Collection and storage
Micromin rals	 Discuss Iron metabolism. Describe Iron deficiency anemia, hemochromatosis, 	2		of urine samples for laboratory analysis, and physical and chemical analysis of urine to detect abnormal constituents.
	 Discuss Iodine and copper. Describe muscle weakness, neurologic defects and abnormal collagen in copper deficiency. Highlight Iodine deficiency and goiter. 	2		Collection and storage of urine samples for laboratory analysis, and
MINERALS	 Discuss Zinc, selenium, chromium, cadmium, manganese, and flouride. Discuss Cardiomyopathy (Keshan disease) in selenium deficiency. Identify growth retardation and impaired wound healing in Zinc deficiency 	1	1	physical and chemical analysis of urine to detect abnormal constituents
NUTRITI ON	 Discuss balanced diet. Define dietary reference intakes, acceptable macronutrient distribution ranges, EAR, RDA, AI and UL. 	2		
	Discuss energy metabolism	2	1	Writing a urine report

	 Define and discuss metabolic rate , factors affecting metabolic rate and basal metabolic rate BMR. Calculate caloric requirement of a 			and interpretation of urine analysis
	 Explain biomedical significance of proteins in nutrition. 	1	1	REVISION
	• Identify biomedical significance of lipids in nutrition.	1		
	• Discuss biomedical significance of carbohydrates in nutrition.	1		
	• Enlist the nutritional requirements in pregnancy, lactation, infancy and old age.	1	1	REVISION.
	• Explain obesity and metabolic syndrome.	1		
Clinical correlation s	 Discuss Protein energy malnutrition (Marasmus and Kwashiorkor). Describe the effects of deficiency of essential fatty acids, anorexia nervosa and bullemia nervosa. Explain how hemorrhoids, chronic constipation and diverticular disease of colon is caused due to low fiber diet. 	3		

TEXTBOOKS AND REFERENCES

- Champe, P.C. & Harvey, E.A. (2017). Biochemistry (Lippincott's Illustrated Reviews), 7nd edition. J.B Lippincort Co.
- 2. Harper's Biochemistry 30th Ed

- 3. Texbook of medical Biochemistry by MN Chatterjea 8th Ed
- 4. Biochemistry by Stryer 7th Ed
- Marks, D.B., Marks, A.D. & Smith, C.M. (2005). Basic Medical Biochemistry: A Clinical Approach. 2nd edition. Williams and Wilkins Co.: Baltimore.
- Robert K. Murray, Daryl K. Granner, Peter A. Mayes, Victor W. Rodwell. (2005). Harper's Biochemistry. 26th edition. McGraw-Hill

TABLE OF SPECIFICATIONS FOR BIOCHEMISTRY

THEORY PAPER FIRST PROFESSIONAL

CONTENTS		SEQs	MCQs
1.	Biochemistry of the cell, cell membrane & membrane	0.5	3
	phenomena. Water, pH & buffers.		
2.	Extracellular matrix	0.5	2

3.	Chemistry of Carbohydrates	1.0	4
4.	Chemistry of Lipids	1.0	5
5.	Chemistry of Proteins and Amino Acids; Plasma	1.0	6
	proteins including immunoglobulins		
6.	Chemistry of Nucleotide and nucleic acids	0.5	3
7.	Enzymes	1.0	5
8.	Vitamins	1.5	7
9.	Nutrition	0.5	2
10.	Minerals & trace elements	0.5	4
11.	Heme metabolism, porphyrins, porphyrias, jaundice,	1.0	4
	hemoglobin & Myoglobin, hemoglobinopathies.		
TOTAL ITEMS		09 SEQs	45 MCQs
TOTAL MARKS		45 Marks	45 Marks

25% of MCQs and SEQs should be clinically oriented or problem- based.

10% marks are allocated for 'Internal Assessment'

Total marks for theory paper: SEQ+ MCQ + Internal Assessment = 45 +45+10=100 Marks

ORAL AND PRACTICAL EXAMINATION FIRST PROFESSIONAL

Oral and practical examination carries 100 marks.

EXAMINATION COMPONENT		MARKS
Α	Internal Assessment	10
В	Practical notebook manual (Internal Examiner)	05
С	Viva voce a) External Examiner: 25 Marks b) Internal Examiner: 25 Marks	50
D	OSPEa)Observed stations (6 Marks)b)Non-Observed Station (16 Marks)	22
Е	Practicala)Principle & supposed calculation: 4 Marksb)Performance: 4 Marksc)Structured Table viva (External Examiner): 5 Marks	13